THE INFLUENCE OF SURFACTANTS ON CASEIN MEMBRANE PREPARATION AND SEPARATION TECHNOLOGIES

Marius Simion¹, Gabriela Paun², Aurelia Meghea³, Jianzhong Ma⁴, Fan Zhang⁵

¹INCD ECOIND, Drumul Podu Dambovitei Street, no 71-73, District 6, Bucharest, Romania, email: mariussic@gmail.com
²INCD SB, 296 Splaiul Independenței, District 6, Bucharest, Romania, email: gpaunroman@gmail.com
³Faculty of Applied Chemistry and Materials Science Department of Inorganic Chemistry, Polizu Street, District 1, Bucharest, Romania, email: a.meghea@gmail.com
⁴Shaanxi Research Institute of Agricultural Products Processing Technology, Xi’an 710021, Shaanxi Province, PR China, e-mail: mjz@sust.edu.cn
⁵College of Resource and Environment, Shaanxi University of Science and Technology, Xi’an 710021, Shaanxi Province, PR China, email: zhangfan213_11@126.com

Abstract

The influence of a classic surfactant: palmitoyl-glycylglycine and bola amphiphilic: bis[2-butyl(sodium bis-thioacetate) sodium dicarboxylate 1,10 decanediyl ester] upon the casein membrane preparation and separation technologies for removing organic pollutants (dyes) was studied by: UV-VIS and FT/IR-ATR spectroscopy, dynamic light scattering, optical and scanning electron microscopy, contact angle, microbiological tests and the separation rates. The membranes were produced by a casting-solvent evaporation technique. The casein powder was dissolved in a water-acetic acid (75:25 v/v) solution with and without plasticizer: glycerol and surfactant (classic or bola), constant continuous stirring for 6-8 hrs. at 40ºC, then degassed the solution for 2 hrs. The solution was poured and afterwards maintained in the oven at 40-50°C for 8-12 hrs. Membranes were obtained with different surfactants which influenced the separation properties of membranes. Ecological membranes are obtained from a biodegradable biopolymer and can be used successfully in removing dyes from wastewaters.

Keywords: bola, surfactants, casein, membranes, dyes

1. Introduction

Casein is the most used natural polymer (collagen, elastin, fibrin, etc.) due to the great variation of products with controlled reactivity and resorption and the physical, mechanical and biological specific properties. As a natural polymer, casein is a completely biodegradable raw material and has certain forming properties to obtains membranes or films, such as good adhesive force and strong resistance high-temperature, water vapor permeability, polishing. However, casein membranes has less extensibility and water resistance due to its structure [1], which could restrict its practical applications. Chemical modification and blending with other materials like surfactants, or including cross-linkers [2] and polymers [3] for the casein have been attempted to correct its deficiencies. There are two types of membranes, namely simple and composite membranes. In this research simple casein membranes are obtained by a uniform casting of the solution with: casein/water- acetic acid/ glycerol/surfactant
on a glass plate. Casein membranes were prepared using or not a plasticizer-glycerol and surfactant (classic or bola) such as palmitoyl-glycylglycine and bola amphiphilic: bis[2-butyl(sodium bis-thioacetate) sodium dicarboxylate 1,10 decanediyl ester]. Although the tensile strength and hydrophobic property improved by introducing a surfactant (classic or bola). The surfactants are also involved in the membrane processes, influencing flow through polymeric porous media, cleaning of membranes during the process and after use or modifying the microstructure of the disperse system for separation. In this work the influence of surfactants upon the microporous structure and retention of some pure water soluble dyes: Yellow III (color index 15985) and Orange III (methyl orange), were studied. The surfactant-dye mixed aqueous solutions obtained by varying the dye and surfactant concentration respectively was characterized by UV-VIS spectroscopy, dynamic light scattering, contact angle. Their environmental applications refer, among others, to the presence of dyes, toxic substances, etc in wastewaters, which became more frequent in the last decades, so adequate solutions for their removal should be found.

2. Experimental
For casein membranes obtained the following materials have been used: Casein was purchased from Zhejiang Huatian Co. Ltd; Glycerol and acetic acid from Sigma-Aldrich. National Institute of Research and Development in Chemical and Petrochemical (ICECHIM from Bucharest, Romania) synthesized [4] and provided:

- palmitoyl-glycylglycine obtained by reaction:

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\begin{align*}
\text{CH}_3-\text{(CH}_2\text{)}_{14}-\text{CO}_2\text{NaOH} & \quad + \quad \text{H}_2\text{O} \quad \text{H}\overline{\text{C}}-\text{CH}_2-\text{CO}-\text{NH} \quad \text{CH}_2\overline{\text{C}}-\text{COOH} \quad + \quad \text{Cl} \\
\text{CH}_3-\text{(CH}_2\text{)}_{14}-\text{CO}_2\text{HCl} & \quad + \quad \text{H}\overline{\text{C}}-\text{CH}_2-\text{CO}-\text{NH} \quad \text{CH}_2\overline{\text{C}}-\text{COOH} 
\end{align*}
\]

- bola amphiphilic: bis [2-butyl (bis-thioacetate, sodium) of sodium dicarboxylate] 1.10 decanediyl ester:

\[
\begin{align*}
\text{NaOOC-CH}_2\text{S} & \quad \text{O} \quad \text{ONa} \\
\text{O} & \quad \text{O} \\
\text{S-CH}_2\text{COONa} & \quad \text{Na} \\
\text{O} & \quad \text{O}
\end{align*}
\]

Ecological membranes formed were used successfully in removing the dyes from wastewaters: Yellow III (food dye, color index 15985) and Orange III (methyl orange) from Merck. The experimental techniques used in this study consist in: UV-VIS and FT/IR-ATR spectroscopy, dynamic light scattering, optical and scanning electron microscopy, contact angle, microbiological tests:

- UV-VIS Spectrophotometry: Perkin Elmer Lambda 25 model (200-800nm range, double beam optical system, 4nm resolution, scanning rate 240nm/min, Abs response type);
- FT/IR-ATR Spectrophotometry: Jasco FT-IR 4200;
- DLS: MALVERN Zetasizer Nano;
- scanning electron microscopy: SEM QUANTA 200 electron microscope;
3. Results and discussion

A new procedure was proposed for obtaining casein membranes by a casting-solvent evaporation technology, and is presented in figure 1:

Casein powder (5 – 5.5%)

Dissolve in a water: acetic acid (75:25 v/v) solution

Introduce or not plasticizer (1 - 2% glycerol)

Introduce or not bola surfactant 2 mg/mL

Constant, continuous stirring for 6-8 hrs

Degassed the solution for 2 hrs

Poured in a Petri plate and was maintained in the oven at 40-50°C for 8-12 hrs

CASEIN MEMBRANE

Fig. 1. Technology for obtaining casein membrane with and without plasticizer: glycerol and surfactant (classic or bola)
These conditions allow the casein molecules from solution to be structured and to form intermolecular bonds without any cross-linking agent. Casein membranes obtained had an microporous structure, fig. 2.

SEM images of the surface casein membranes gives information on the surface morphology of the unused membranes. The surface of the casein membranes consists of pores of varying size. Cross-sections of the membranes were prepared to assess internal structure. Cross-sections were cut using a scalpel and fractured. Compression of the structure is visible in Fig. 3. Casein membrane consists of a finger-like microsubstructure.

Optical microscopy of membranes confirms microstructure observed by electron microscopy. FTIR-ATR spectroscopy of casein membranes showed characteristic peaks for each type of membrane with or without plasticizer and surfactant (classic or bola). Microbiological tests were performed according to European standard using Sabouraud culture medium inoculated with Staphylococcus Aureus, placed over the samples. The strongest antibacterial effect was found for the membrane with bola surfactant.

Casein membranes interactions with surfactants-dyes and separation rates of dyes were studied with: UV-VIS, dynamic light scattering, scanning electron microscopy, contact angle. To study dyes-surfactant interactions, the following were chosen: b)-casein membrane with plasticizer, c)-casein membrane with plasticizer and bola surfactant. The micellar adsorption wavelength from 470 nm, characteristic for Orange III solution, Yellow III at 430 nm. UV-VIS spectroscopy showed that in the absence of bola surfactant, in the plot of the normalized flux vs. the solution concentration, the Yellow III dye exhibits an extreme point at a concentration of 0.14 g/dL, both for casein membranes with or without bola surfactant. The Yellow III dye also shows a maximum
retention for small concentrations (0.01 g/dL). At greater concentrations, the retention on the membrane-b) is zero, respectively 40% for the membrane-c).

The maximum point can be assigned to an adsorption phenomenon on the membrane, but the membrane-c) exhibits a real retention, Fig. 4. In the case of Orange III dye the membrane-b) is fouled and hence the normalized flux is decreasing, fig.5.

![Graph of normalized flux and retention vs concentration](image1)

The presence of the bola surfactant in the dye solution leads to the formation of some complexes (associates), [5]. These complexes may be swelled with an organic compound, or may be caused by the electrostatic interactions, favouring the increase of the retention both for the Orange III dye and the bola surfactant, fig.6.

![SEM image of associates](image2)

The membrane-c) obtained with bola surfactant exhibits lower contact angle towards water than the membrane-b) without surfactant, 120° to 93°.

4. Conclusions

A variety of techniques for microstructural analysis were used to study the surface morphology and internal microstructure of casein membranes [6]. Comparison of the methods shows that either optical or SEM microscopy examination showed unused flat sheet casein membranes.

The presence of surfactants in the composition of the membranes formed was emphasized in order to maintain the hology and membrane performances. Surfactants
in the casting solution alters the size, as well as the density of pores and the roughness of casein membranes surface. The surfactants analysed, yield membranes with small and dense pores and with smooth surface. Also, surfactants in the dyes aqueous solution influenced the separation rates. Ecological membranes are obtained from a biodegradable biopolymer and can be used successfully in removing dyes from wastewaters. The actual European Community strategy related to maintenance of health of population, quality of life, and of the environment encourages the new technologies for pollution abatement.

References