

METAL – TITANIUM DIOXIDE DOPED CATALYSTS FOR WASTEWATER TREATMENT UNDER SIMULATED SOLAR LIGHT

Lucian-Alexandru Constantin, Ioana Alexandra Ionescu, Mirela Alina Constantin, Mihai Stefanescu, Nicoleta Mirela Marin

National Research and Development Institute for Industrial Ecology – ECOIND, 57-73 Drumul Podu Dambovitei St., Bucharest – 6, Romania

Introduction

TiO₂ assisted photo catalysis was widely studied in the last years for the advanced degradation of organic compounds from aqueous systems, mainly for those not removed by conventional treatment processes. The main drawback of TiO₂ is that it is activated only by UV radiation and uses only about 5% of available solar radiation. TiO₂ doping with metals proved to be a good method to improve its photocatalytic properties and to enhance its response to solar radiation.

Materials and methods

Three types of metal-doped TiO₂ catalysts (Fe-TiO₂, Ni-TiO₂, Co-TiO₂) were prepared using sol-gel method. The TiO₂ precursor was titanium (IV) isopropoxide and Fe(NO₃)₃ · 9 H₂O, Ni(NO₃)₂ · 6 H₂O, Co(NO₃)₂ · 6 H₂O (were used as metal sources. Catalysts with 1wt%, 2wt%, 5wt% were prepared. A lamp with the following characteristics: 35W, 380-800 nm, 1000 lumens, photosynthetic photon flux density 300 μmol quanta/m²·s was used to simulate solar radiation. Dimensional analysis was performed using a Mastersizer 2000 – Malvern equipment and a FEI Quanta FEG 250 scanning electronic microscope was used for morphological characterisation and EDX characterisation. Two sets of experiments were performed using a synthetic solution of methylene blue – MB and real wastewater from a municipal wastewater treatment plant.



Experimental setup

Results and Conclusions

Dimensional analyse

Catalyst	d (0.1) μm	d (0.5) μm	d (0.9) μm
1% wt. Ni-TiO ₂	0.176	0.331	0.634
1% wt. Co-TiO ₂	0.176	0.330	0.630
TiO ₂ Merck	0.175	0.327	0.612
1% wt. Fe-TiO ₂	0.388	0.505	0.979
P25 Degussa	0.295	0.459	1.017

EDX results

1 wt% Fe-TiO ₂		1 wt% Ni-TiO ₂		1 wt% Co-TiO ₂	
Element	Atomic %	Element	Atomic %	Element	Atomic %
O	66.43	O	66.97	O	70.53
Fe	0.22	Ni	0.21	Co	0.90
Ti	33.35	Ti	32.82	Ti	28.57

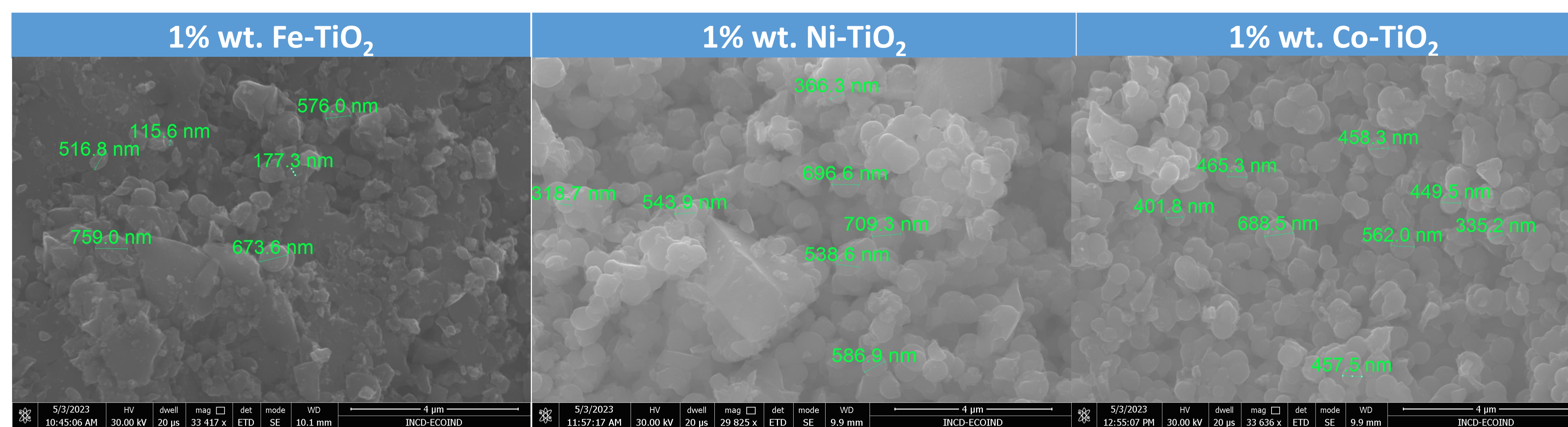
Photocatalytic activity of all catalysts were tested using MB. MB initial concentration and after 3 hours of exposure to solar simulated radiation and a catalyst dose of 100 mg/L were measured based on the absorbance at 662 nm (maximum MB absorbance).

Best results were obtained for 2wt% Metal – TiO₂ catalysts with a MB degradation efficiency of 6.00% for 2wt% Co-TiO₂, 6.91% for 2wt% Ni-TiO₂ and 13.82% for 2wt% Fe-TiO₂.

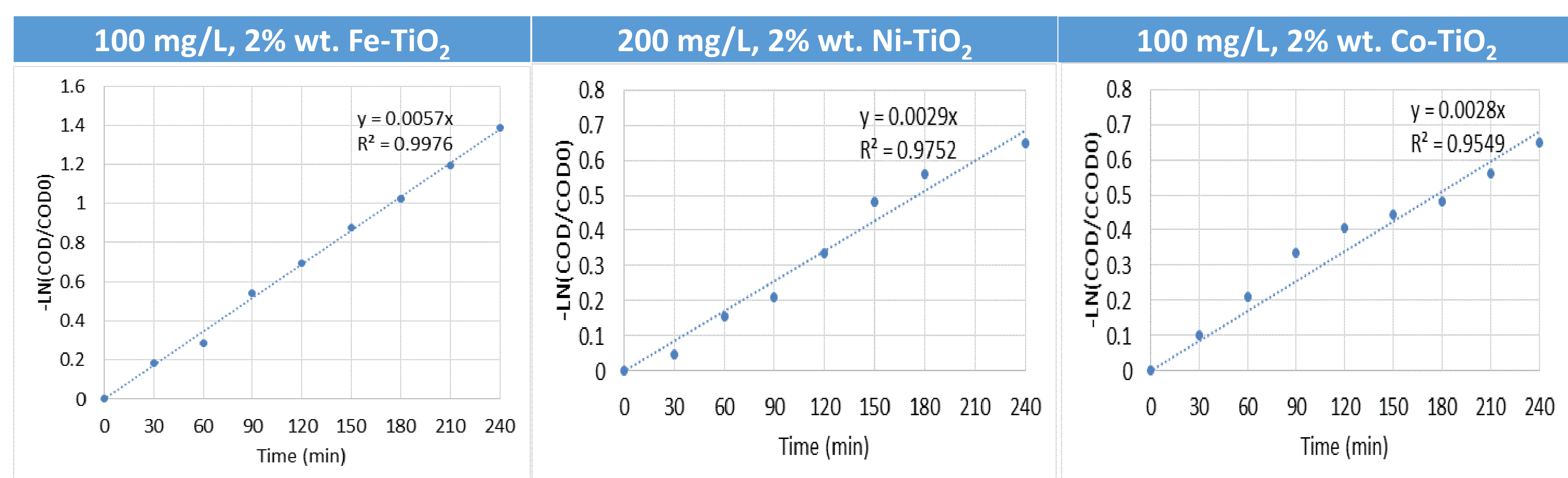
The 2wt% Metal-TiO₂ catalysts were further used for treatment of real wastewater samples varying the initial catalyst concentration.

The best degradation efficiencies for organic compounds (expressed as COD) after 120 minutes of irradiation were obtained for the following catalysts doses: 200 mg/L for 2wt% Ni-TiO₂ (28.57%), 100 mg/L for 2wt% Co-TiO₂ (33.33%), 100 mg/L for 2wt% Fe-TiO₂ (50.00%).

Further increase of irradiation time to 240 minutes led to a degradation of organic compounds (expressed as COD) of 47.69% for both 2wt% Ni-TiO₂ and 2wt% Co-TiO₂ and 75.00% for 2wt% Fe-TiO₂.



SEM images



Pseudo-first order kinetic for optimum catalysts, irradiation time 240 min

Acknowledgments

This work was carried out through the "Nucleu" Program within the National Research Development and Innovation Plan 2022-2027 with the support of Romanian Ministry of Research, Innovation and Digitalization, contract no. 3N/2022, Project code PN 23 22 03 01.